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Buxar (Buxar). Notes for  
B.Sc part 3<sup>rd</sup>, paper VI, Unit  
= 2(3).

Question :- Write Notes on structure  
of Amino Acid?

Answer :-

Amino acid :- Amino acid are  
organic compounds that contain  
amine ( $-NH_2$ ) and carboxyl ( $-COOH$ )  
functional groups, along with  
a side chain (R group) specific  
to each amino acid. The key  
elements of an amino acid  
are carbon (C), hydrogen (H),  
oxygen (O), and nitrogen (N),  
although other elements are  
found in the side chains  
of certain amino acids. About  
500 naturally occurring  
amino acids are known  
(though only 20 appear in  
the genetic code) and can  
be classified in many ways.

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They can be classified according to the core structural functional group's locations as alpha- ( $\alpha$ -), beta- ( $\beta$ -), gamma- ( $\gamma$ -) or delta- ( $\delta$ -) amino acids; other categories relate to polarity, pH level, and side chain group type aliphatic, acyclic, aromatic, containing hydroxyl or sulfur, etc. In the form of proteins, amino acid residues form the second - largest component (water is the largest) of human muscles and other tissues.

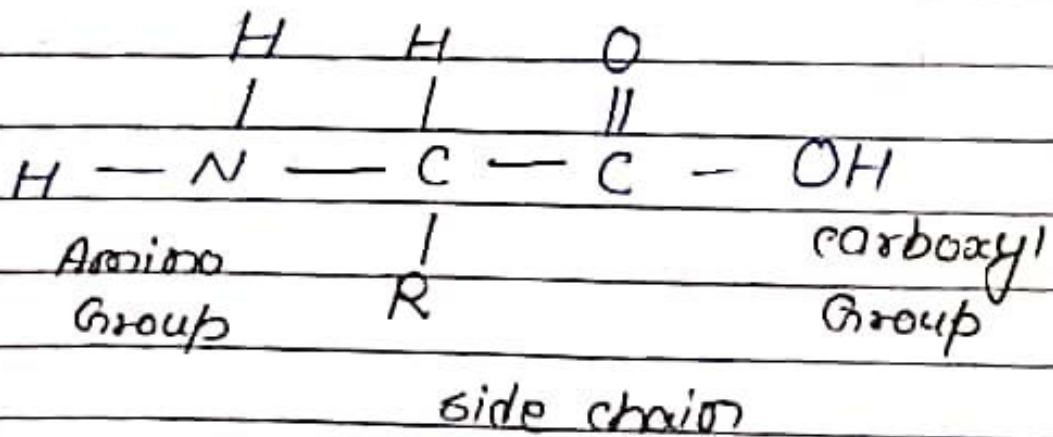
Beyond their role as residues in proteins, amino acids participate in a number of processes such as neurotransmitter transport and biosynthesis.

Nine proteinogenic amino acids are called "essential" for humans because they cannot be produced from other compounds by the human body and so must be taken in as food. Others may be conditionally essential for certain ages or

medical conditions. Essential amino acids may also differ between species.

Because of their biological significance, amino acids are important in nutrition and are commonly used in nutritional supplements, fertilizers, food, and food technology. Industrial uses include the production of drugs, biodegradable plastics and chiral catalysts.

### Amino Acid Structure :-



### Amino Acid

### side chains :-

Amino acids are designated as  $\alpha$ - when



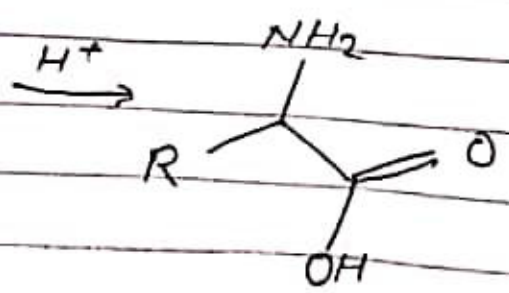
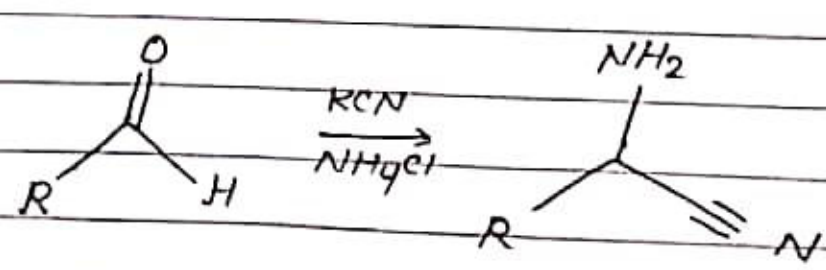
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In aqueous solution amino acids exist in two forms as illustrated at the right. the molecule form and the zwitterion form in equilibrium with each other.

The two forms co-exist over the pH range  $pK_1 - 2$  to  $pK_2 + 2$ , which for glycine is pH 0-12. The ratio of the concentration of the two isomers is independent of pH. The value of this ratio cannot be determined experimentally.

Synthesis :-



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## Chemical synthesis :-

The commercial production of amino acids usually relies on mutant bacteria that overproduce individual amino acids using glucose as a carbon source. Some amino acids are produced by enzymatic conversions of synthetic intermediates. e.g. Aminothiazoline-4-carboxylic acid is an intermediate in the industrial synthesis of L-cysteine. For example, Aspartic acid is produced by the addition of ammonia to fumarate using Lyase.

## See also :-

- Amino acid dating
- Beta peptide
- Degman
- Epsilon
- Homochirality
- Hypoaminoacidemia
- Leucines
- Miller - Urey experiment
- Nucleic acid sequence